Guest(Solvent)-Dependent Enantioselective Crystallization of 7-Bromo-1,4,8-triphenyl-2,3-benzol3.3.0locta-2,4,7-trien-6-one As the Inclusion Complex

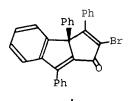
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Abstract: The optical resolution of the title compound is easily achieved by its preferential crystallization as an inclusion complex with the solvent used for the crystallization.

The title compound (1)¹ has been found to include a wide variety of solvent as guests and form crystalline inclusion crystals. In the inclusion crystallizations, racemates or conglomerates of 1 were formed depending on the choice of solvent. In the latter case, the inclusion crystals consisting of one enantiomer of 1 were formed preferentially and the optical resolution of 1 could be performed.

Recrystallization of rac 1 from the solvents shown in Table 1 gave a 1:1 inclusion crystal of the rac 1 with the solvent as yellow crystals (Table 1). On the other hand, recrystallization of the rac 1 from the solvents shown in Table 2 gave 1:1 inclusion crystals of the optically active 1 with the solvent as yellow crystals (Table 2). For example, recrystallization of the rac 1 (mp 185-186 °C) from THF gave a 1:1 inclusion complex of the optically active 1 and THF, each single crystal consisting of optically pure (+)- or (-) -1. By seeding with one crystal of optically pure 1 during the recrystallization of rac 1, a large quantity of the optically pure 1 could be obtained.



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| Rac Compound of 1 with Solvent (Guest)" | | | | | | | | |
|---|-------------------------|--------------------|-------------------|---------|---------------------------------|--|--|--|
| Solvent(Guest) | Mp/°C | Photo- chromism | Solvent(Guest) | Mp/°C | Photo- chromism ^b | | | |
| VOL Me | 112-115 | + | Me | 100-102 | + | | | |
| \diamond | nd^{c} | + | PhNH ₂ | 59-61 | | | | |
| $\langle \rangle$ | nd | | N N | 145-149 | | | | |
| O | 109-112 | + | Me2N-NMe2 | 177-179 | - | | | |
| CHC13 | 101-105 | + | Me0-OMe | 159-165 | + | | | |
| CHBr ₃ | 129-135 | | PhCl | 88-90 | + | | | |
| CH2C12 | nd | + | i nei | 00-90 | ſ | | | |
| PhOMe | 127-131 | + | clCl | 132-138 | -+ | | | |

Table 1. Formation of 1:1 Inclusion Crystals of

^aAll inclusion compounds are formed as yellow crystals. ^bVarious colors appear depending on a kind of the solvent included. ^cNot distinct.

For example, to a solution of the rac 1 (10 g) in THF (50 ml) was added one piece of the crystal of the (+)-1·THF complex and the mixture kept at room temperature for 12 h to give the (+)-1·THF complex, after one recrystallization from THF, 0.81 g (14% yield, 100% ee, $[\alpha]_D^{20}$ +479.8 (*c* 0.21 in CHCl₃)). Distillation of THF from the complex in vacuo gave (+)-1 of 100% ee (0.7 g, 14% yield, mp 234-236 °C, $[\alpha]_D^{20}$ +480 (*c* 0.41 in CHCl₃)). To the filtrate left after filtration of the crude (+)-1·THF complex, one piece of the crystal of the (-)-1·THF complex was added and the solution was kept at room temperature for 12 h to give the (-)-1·THF complex, after two recrystallizations from THF, 1.1 g (19% yield, 100% ee, $[\alpha]_D^{20}$ -479.5 (*c* 0.20 in CHCl₃)). Distillation of THF from the complex in vacuo gave (-)-1 of 100% ee (0.96 g, 19% yield, mp 234-236 °C, $[\alpha]_D^{20}$ -480 (*c* 0.41 in CHCl₃)). Optical purity of the (+)- and (-)-1 was determined by HPLC using a column containing an optically active solid phase, YMC A-K03.

Some inclusion crystals summarized in Tables 1 and 2 showed photochromism and various colors appeared depending on a kind of solvent included. X-Ray crystal structural studies of the crystals which show this photochromism will be reported in the near future.

| Rac Mixture of 1 with Solvent (Guest) | | | | | | | |
|--|----------|---------------------------------|------------------|---------|---------------------------------|--|--|
| Solvent(Guest) | Mp/°C | Photo- chromism ^b | Solvent(Guest) | Mp/°C | Photo- chromism ^b | | |
| $\langle \rangle$ | 112-114 | | | 135-139 | + | | |
| ∠o Me 2 | nc^{c} | | Me | 98-102 | t | | |
| \bigcirc | 120-125 | + | N-Me | 119-123 | - | | |
| Cot Me 3 | nd | - | Mel | nd | | | |
| $\left(\begin{array}{c} \circ \\ \circ \end{array} \right)$ | 122-128 | + | EtBr | na | | | |
| \sim | _ | | Et1 | na | | | |
| L_H | nd | | CC14 | 127-130 | + | | |
| | nd | | CBr ₄ | 129-134 | + | | |

Table 2. Formation of 1:1 Inclusion Crystals of Pac Mixture of 1 with Solvent (Cuest)^a

^aAll inclusion compounds are formed as yellow crystals. ^bVarious colors appear depending on a kind of the solvent included. ^cNot distinct.

It is interesting to note that the solvents which form complexes with rac 1 do not form complexes with optically active **1 and vice versa**. For example, the powdered (+)-1 obtained by the evaporation of THF from the (+)-1·THF complex turns to guest-free crystals by recrystallization from benzene. The difference of the role between the solvents shown in Tables 1 and 2 will be clarified by X-ray crystal structural studies of the inclusion complexes.

By using optically active **1** for the complexation, some guest compounds could be resolved. For example, when a solution of (-)-1 (1.3 g) in rac 2-methyltetrahydrofuran (2) (5 g) was kept at room temperature for 12 h, a 1:1 inclusion complex of (-)-1 and (-)-2 was formed as yellow prisms (1.32 g), which upon distillation gave $(R)-(-)-2^3$ of 32% ee (0.1 g, 4% yield, $[\alpha]_D^{20}$ -6.4 (c 0.96 in CHCl₃)). The optical purity was determined. by comparison of the $[\alpha]_D$ value with that of an authentic sample.³ By the same method, 2-methylpyran (3) was resolved by (-)-1 to give $(R)-(-)-3^3$ of 30% ee in 8% yield. Although the efficiency of the resolution is not high, chiral recognition between 1 and 2 or 3, which have no binding groups such as hydroxyl for hydrogen bonding, is interesting.

Further interesting enantioselective complexation between 1 and 3 has been observed. When a single crystal of $(+)-1\cdot(+)$ -3 complex was used to seed at the recrystallization of rac 1 (2.0 g) from rac 3 (20 g), inclusion crystals of (+)-1 of 99.0% ee and (+)-3 of 29.4% ee were obtained (0.22 g, 18.6% yield).

Acknowledgement

We wish to thank for the Ministry of Education, Science and Culture, Japan, for Grant-in-Aid for Scientific Research on Priority Areas, No. 63840017.

References and Notes

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- YMC A-K03 is available from Yamamura Chemical Laboratories Co. Ltd., Kyoto, Japan.
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